

pressure interval 5 to 45 kb. In all cases both samples showed a polymorphic transition at a pressure corresponding to  $P_{tr}$  found for pure KCl at the same corresponding temperature (5). Furthermore, the width of this transition pressure interval ( $P_w$ ) (5, 6) in these mixtures was approximately the same as  $P_w$  measured for pure KCl (5). The transition volume ( $\Delta V_{tr}$ ) for these compositions was found to be approximately 90% and 75%, respectively, of  $\Delta V_{tr}$  measured for pure KCl at the same corresponding temperature (5).

#### B. KCl-NaCl

Mixtures with compositions between 5 and 95 m/o KCl were examined. Data for the mean transition pressure ( $P_{tr}^m$ ) (6) found for the polymorphic phase change in binary salts in this system at isotherms 300 through 800° are shown in Fig. 1. The

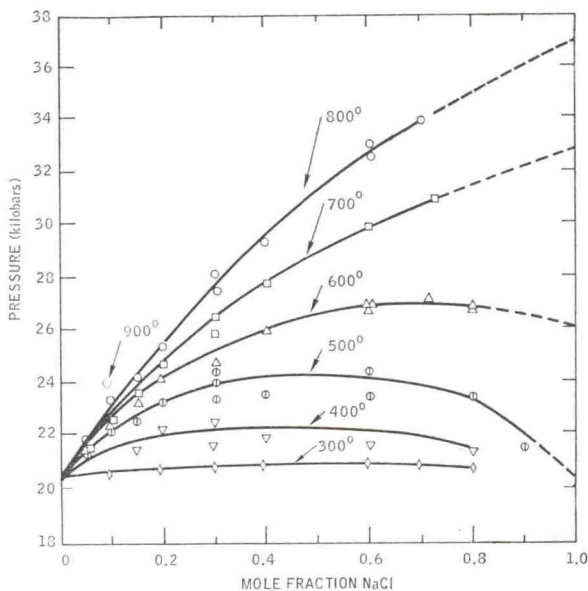


FIG. 1. Polymorphic transition pressure ( $P_{tr}^m$ ) of salts in the KCl-NaCl system in the temperature range 300–900°.

width of the polymorphic transition pressure interval ( $P_w$ ) measured for salts in this system is shown for two temperatures, 500 and 800°, in Fig. 2. The volume change which occurs during the polymorphic transition in these salts was also measured over the temperature interval 300° to 800° at 100° intervals. These transition volume data are shown in Fig. 3. Below 300° the transition of these mixtures is so sluggish and hysteresis so large that no reliable and reproducible extrapolation could be made from the pre- and post-transition  $P$ - $V$  curves; thus no  $P_{tr}$  and  $\Delta V_{tr}$  data are given for temperatures below 300°. From Fig. 3 it can be seen that the phase transition volume ( $\Delta V_{tr}$ ) becomes very small at high temperatures and at high mole fractions of NaCl. The highest composition of NaCl (95 m/o) for which a transition volume

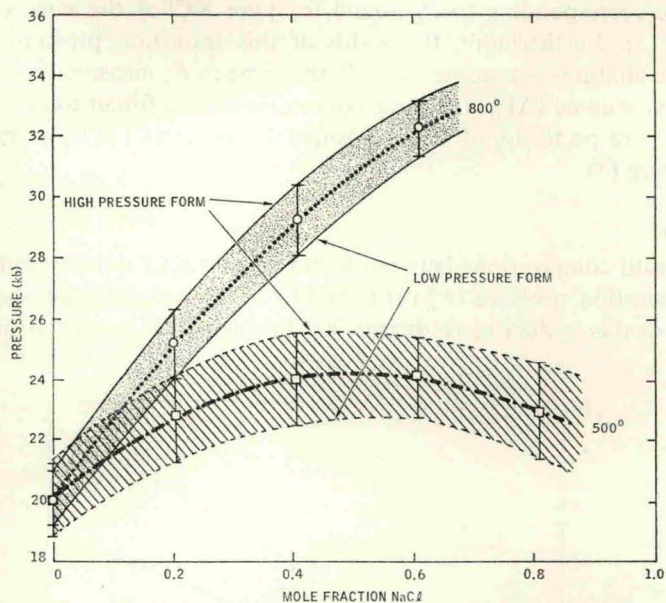


FIG. 2. Polymorphic transition pressure of salts in the KCl-NaCl system showing width of the transition pressure interval ( $P_w$ ) at 500° and 800°.

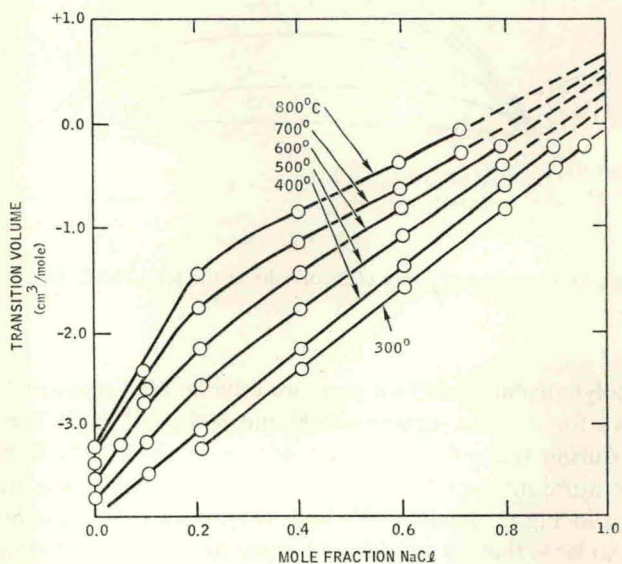


FIG. 3. Transition volume ( $\Delta V_{tr}$ ) of the pressure-induced phase change for salts in the KCl-NaCl system at the isotherms 300°-800°.